

# Development of a novel bipolar stacked Iron/Air Battery Stack

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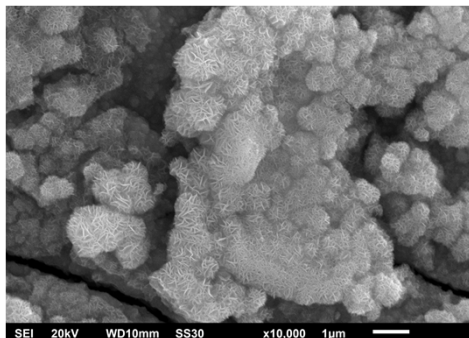
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## MOTIVATION

Electrical energy from renewable sources needs to be buffered due to high fluctuations. The iron/air battery is a promising system for this application. Iron is a cheap, abundant, and environmentally safe material. The theoretical energy density of this system is about six times higher compared to state of the art Li-Ion technology [1,2].

### IRON ELECTRODE

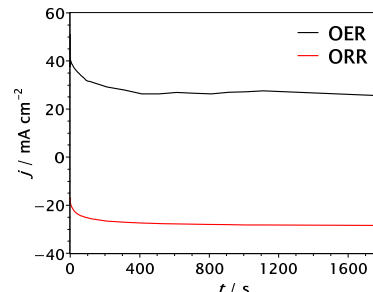
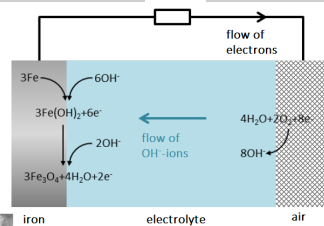
- Traditional Iron electrodes are hot pressed and have a low coulombic efficiency
- Goal: electrochemical prepared iron electrodes and CE >80 % with electrode and electrolyte additives
- Current state: high capacities  $\approx 1000 \text{ mAhg}^{-1}$
- Synthesis of hybrid Fe-Bi-S-Electrodes



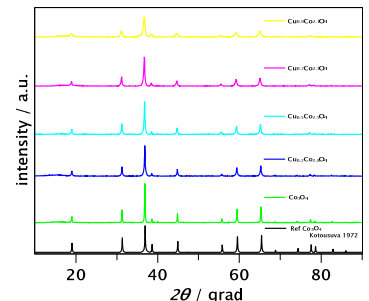
SEM Image of Fe-Bi-hybrid

### AIR ELECTRODE

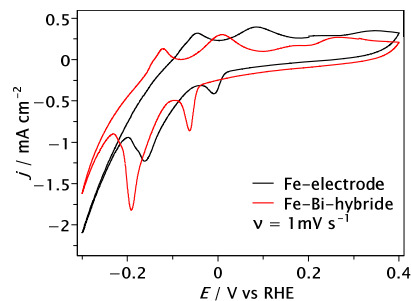
- Air Electrodes are carbon based and utilize noble metal catalyst materials
- Carbon corrodes at low voltages ( $\sim 0.8 \text{ V}$ ) and is not applicable [3]
- Goal: Carbon and noble metal free air electrode. Transition metal oxides with spinel structure (especially  $\text{Co}_3\text{O}_4$ ) show high bifunctional activity [4]



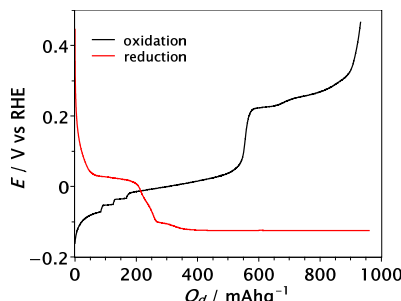
Chronoamperometry of air electrode with commercial  $\text{Co}_3\text{O}_4$  catalyst at 0.3 and 1.8V in 1 M KOH @ 100  $\text{ml min}^{-1}$  air flow



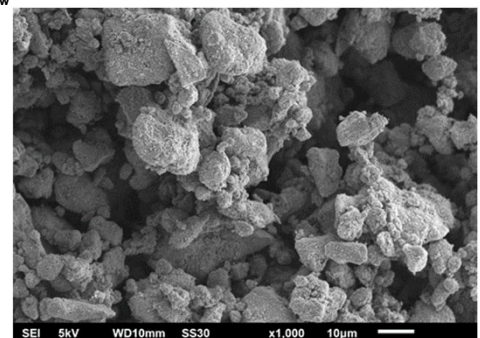
XRD data of copper substituted  $\text{Co}_3\text{O}_4$



Comparison Fe- and Fe-Bi-hybrid electrode pH = 14.9



Fe electrode reduction & oxidation curve pH = 14.9 @ C/4



SEM image of in-house made  $\text{Co}_3\text{O}_4$

## CONCLUSION AND OUTLOOK

- ✓ High capacity iron electrodes have been produced
- ✓ Fe-Bi Hybrid electrode successfully produced
- Enhancing cyclic stability by improving morphology
- Increasing coulomb efficiency by using additives
- 10 cell battery stack with high capacity and high cycle stability
- Bipolar design consisting of polymer/carbon based bipolar plates

- ✓ Air electrodes with high initial current densities and commercial  $\text{Co}_3\text{O}_4$  catalyst have been produced
- Study the dependence of element substitution and synthesis route of  $\text{Co}_3\text{O}_4$  and other spinel structures on electro-catalytic activity
- Air electrodes with improved, in-house made, catalyst
- Parameter study to find optimum material composition and process parameters

### References

- [1] Tarascon J-M, Armand M (2001) Issues and challenges facing rechargeable lithium batteries. *Nature* 414(6861):359-367  
 [2] Y. Li and J. Lu, "Metal-Air Batteries: Will They Be the Future Electrochemical Energy Storage Device of Choice?," *ACS Energy Letters*, vol. 2, no. 6, pp. 1370-1377, 2017, doi: 10.1021/acsenenergylet.7b00119.

[3] T. R. Crompton, *Battery reference book*, 3rd ed. Oxford: Newnes, 2007.

[4] D. Wittmaier et al., Bifunctional, Carbon-Free Nickel/Cobalt-Oxide Cathodes for Lithium-Air Batteries with an Aqueous Alkaline Electrolyte, *Electrochimica Acta*, vol. 149, pp. 355-363, 2014